VALPARAISO UNIVERSITY

Abstract

Pd(0)-catalyzed trimethylenemethane (TMM) cycloaddition reactions have been used extensively to generate disubstituted 5membered rings with high levels of regioselectivity, chemoselectivity, and stereoselectivity. We aim to improve upon existing methodologies by introducing an epoxide into the TMM precursor, which should yield more highly functionalized products. A three step synthesis of a TMM precursor has been achieved in 20% overall yield. Efforts are underway to determine the reactivity of this precursor in Pd⁰-catalyzed cycloaddition reactions with substituted alkenes.

Background

• Natural products and pharmaceutical drug targets often contain rings of varying size and complexity.¹⁻³ Having methods for stereoselectively and regioselectively producing these rings is crucial to organic synthesis.







-)-Rocaglamide *Figure 1. Compounds synthesized using Pd*⁰*-catalyzed TMM [3+2] cycloadditions.*

• The Trost group at Stanford has studied Pd⁰-catalyzed TMM [3+2] cycloadditions rather extensively, successfully generating disubstituted cyclopentane structures.⁴



Figure 2. General mechanism of cycloadditions performed by the Trost group.

• Pd⁰-catalyzed π -allyl alkylations have been done successfully on epoxides⁵, but this approach has not yet been used in TMM cycloadditions.



Figure 3. Pd^0 -catalyzed π -allyl alkylation mechanism with TMM epoxide precursor.

• We hypothesize that an epoxide-containing TMM precursor will go through a similar mechanistic pathway as in Figure 3 to yield a trisubstituted 5-membered ring, shown in Figure 4.



Figure 4. Proposed cycloaddition mechanism with epoxide TMM precursor.

Development of a Novel Epoxide-Containing Trimethylenemethane **Precursor for Palladium-Catalyzed Cycloadditions**

Mara L. Paterson and Dr. Kevin Jantzi*

Department of Chemistry, Valparaiso University





Optimization of TMM Precursor Synthesis



Scheme 1. Original synthesis of epoxide precursor.

- The preliminary focus of this research was to further optimize the synthesis shown in Scheme 1 that had been used previously in this project. The primary goals were time and cost efficiency, as well as minimizing the amount of hazardous waste.
- Optimization was guided by the goal for this synthesis to ultimately be used by an individual student in an undergraduate research laboratory.



Scheme 2. Optimized synthesis of epoxide precursor.

- Drying diethyl ether with Na⁰ for the first synthetic step was determined to be unnecessary, as it improved the yield by a negligible amount when compared to ether that had not been dried.
- MnO₂ was found to be the best oxidant due to mild conditions, simple workup, and a reduced amount of hazardous waste.⁶
- A large excess (13-17 eq.) of MnO₂ used in each oxidation, but 10-12 eq. of activated MnO₂ was recovered from each reaction and could be used again in future oxidations.

Table 1. Optimization of carbocupration, producing 1.			Table 2. Optimization of the oxidation of 1 to 2.				
Solvent	TMSCH ₂ MgCl	Yield		Solvent	Time	Temperature	Yield
Dry THF	1 equivalent	43%		Pentane	2 h	RT	0%
				Pentane	24 h	RT	12%
Dry THF	1.5 equivalents	36%		CH_2Cl_2	2 h	RT	12%
Dry ether Ether	1 equivalent 1 equivalent	39% 40%		CH_2Cl_2	2 h	Reflux	43%
				CH_2Cl_2	24 h	RT	68%
				CH_2Cl_2	24 h	Reflux	11%

• A major benefit of this synthetic approach is that while the final epoxide precursor needed to be purified via distillation, no purification steps were required for the intermediate products, as shown by the ¹H NMR spectra in Figures 5 and 6 below.



Figure 5. ¹H NMR spectrum of Compound 1.



Figure 6. ¹H NMR spectrum of Compound 2.

	Pd ⁰
• Our ep investig alkenes.	oxide- ated f
<i>Figure 7. G</i> • Alkenes reaction	TM eneral pr were s perfe
precurso O C chalco	or. /
Figure 8. A	lkenes cu
 Continu to detern Incorpor effects c Synthes 	e scree nine if rate R on regio ize the
cycload	dition 1
 TMS	R
Figure 9. St without add	ıbstitutec itional sı
 Trost, B. 1986, <i>10</i> Paquette 	M.; L 8, 284-2 , L. A.;
M.; Ande 3. Trost, B.	erson, L M.; Gre
1990 , <i>11</i> . 4. Trost, E	2, 9022 3. M.;
<i>Soc.</i> 201 5. Trost, B.	1 , <i>43</i> (1 M.; M
5972. 6. Tojo, G	.; Fern
<i>Ketones</i> ; 7. Trost, B.	Spring M.; Ch
Valparative	so Uni Work



-Catalyzed Cycloadditions

-containing TMM precursor is currently being for its reactivity with various electron-deficient



roposed cycloaddition reaction with our novel TMM precursor.

selected based on previous success in cycloaddition formed using the model acetate-containing TMM



benzylideneacetone dimethyl maleate methyl cinnamate

urrently being tested for reactivity in [3+2] cycloadditions.

Future Work

ening alkenes and cycloaddition reaction conditions the predicted products are being formed

groups into TMM precursor to determine the oselectivity and stereoselectivity

e aziridine equivalent TMM precursors and repeat reactions and analysis







d epoxide TMM precursor and aziridine TMM precursors with and ubstitution.

References

Lynch, J.; Renaut, P.; Steinman, D. H. J. Am. Chem. Soc. 291.

Sauer, D. R.; Cleary, D. G.; Kinsella, M. A.; Blackwell, C. L. G. J. Am. Chem. Soc. 1992, 114, 7375-7387.

eenspan, P.D.; Yang, B.V.; Saulnier, M.G. J Am Chem Soc. -9024.

Silverman, S. M.; Stambuli, J. P. J. Am. Chem. 9), 19483–19497.

Iolander, G. A. J. Am. Chem. Soc. 1981, 103 (19), 5969-

nández Marcos. Oxidation of Alcohols to Aldehydes and er: New York, 2010.

nan, D. M. T. J. Am. Chem. Soc. 1979, 101, 6429-6431.

Acknowledgements

iversity Department of Chemistry Creative Work and Research Committee Scholarship and Advising Committee • Dr. Jeff Pruet, Dr. Tom Goyne, Dr. Haiying He